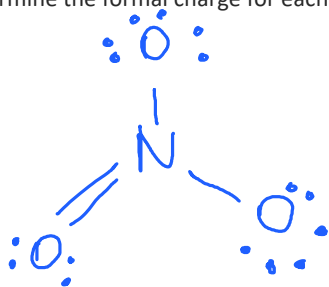


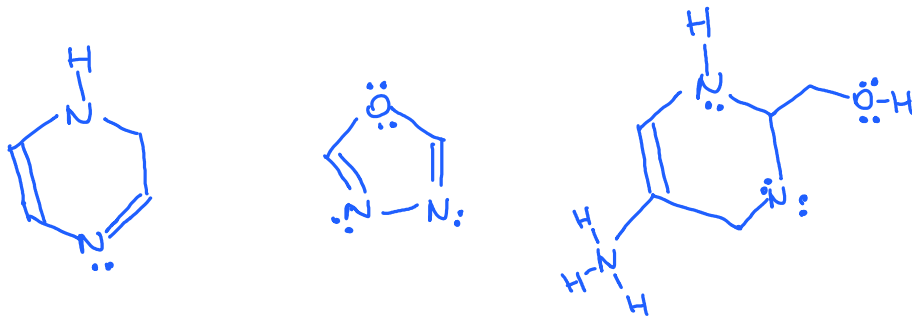
FORMAL CHARGES PRACTICE

FORMAL CHARGE = NUMBER OF VALENCE ELECTRONS - (NUMBER OF BONDS + NUMBER OF LONE ELECTRONS)

1. Determine the formal charge for each atom



2. Determine the formal charge of each nitrogen atom and oxygen atom



3. Determine any formal charges that are present

